

# Benzenamine, 4,4',4''-methylidynetris[N,N-dimethyl-

Other names:

Aniline, 4,4',4''-methylidynetris[N,N-dimethyl-

C.I. Basic Violet 3, leuco

Leucocrystal Violet

Leucomethyl green

Tris[p-(dimethylamino)phenyl]methane

Tris[4-(dimethylamino)phenyl]methane

4,4',4''-(Hexamethyltriamino)triphenylmethane

4,4',4''-Methylidynetris(N,N-dimethylaniline)

4,4',4''-Tris(dimethylamino)triphenylmethane

4,4',4''-Methylidynetris-(N,N-dimethylaniline)

4,4',4''-Tris(N,N-dimethylaminophenyl)methane

p,p',p''-Methylidynetris(N,N-dimethylaniline)

Leucogentian violet

NSC 7338

N,N,N',N',N'',N''-hexamethyl-4,4',4''-methylidynetrianiiline

**Inchi:** InChI=1S/C25H31N3/c1-26(2)22-13-7-19(8-14-22)25(20-9-15-23(16-10-20)27(3)4)21-11

**InchiKey:** OAZWDJGLIYNYMU-UHFFFAOYSA-N

**Formula:** C25H31N3

**SMILES:** CN(C)c1ccc(C(c2ccc(N(C)C)cc2)c2ccc(N(C)C)cc2)cc1

**Mol. weight [g/mol]:** 373.53

**CAS:** 603-48-5

## Physical Properties

Property code	Value	Unit	Source
gf	797.86	kJ/mol	Joback Method
hf	313.16	kJ/mol	Joback Method
hfus	47.00	kJ/mol	Joback Method
hvap	85.80	kJ/mol	Joback Method
log10ws	-4.98		Crippen Method
logp	5.065		Crippen Method
mcvol	321.770	ml/mol	McGowan Method
pc	1404.84	kPa	Joback Method
tb	903.26	K	Joback Method
tc	1135.73	K	Joback Method
tf	570.74	K	Joback Method
vc	1.159	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1016.38	J/mol×K	903.26	Joback Method
cpg	1034.43	J/mol×K	942.00	Joback Method
cpg	1051.16	J/mol×K	980.75	Joback Method
cpg	1066.72	J/mol×K	1019.49	Joback Method
cpg	1081.23	J/mol×K	1058.24	Joback Method
cpg	1094.84	J/mol×K	1096.98	Joback Method
cpg	1107.68	J/mol×K	1135.73	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C603485&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C603485&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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