

# 2-Aminopent-4-enoic acid, N-(2-ethylhexyloxycarbonyl)-, undecyl ester

Inchi:	InChI=1S/C25H47NO4/c1-5-9-11-12-13-14-15-16-17-20-29-24(27)23(18-7-3)26-25(28)3
InchiKey:	ILSSPMOEAQIIEZ-UHFFFAOYSA-N
Formula:	C25H47NO4
SMILES:	C=CCC(N=C(O)OCC(CC)CCCC)C(=O)OCCCCCCCCCCC
Mol. weight [g/mol]:	425.64

## Physical Properties

Property code	Value	Unit	Source
hf	-901.28	kJ/mol	Joback Method
hvap	101.44	kJ/mol	Joback Method
log10ws	-7.44		Crippen Method
logp	7.152		Crippen Method
mcvol	383.670	ml/mol	McGowan Method
pc	798.43	kPa	Joback Method
rinpol	2766.00		NIST Webbook
rinpol	2766.00		NIST Webbook
tb	1034.65	K	Joback Method
tc	1283.08	K	Joback Method

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U393166&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U393166&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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