

# (Z)-Pentacos-16-ene-2,4-dione

<b>Inchi:</b>	InChI=1S/C25H46O2/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-25(27)2
<b>InchiKey:</b>	MDFRVEWZNNVTRI-KHPPLWFESA-N
<b>Formula:</b>	C25H46O2
<b>SMILES:</b>	CCCCCCCC=CCCCCCCCCCCCC(=O)CC(C)=O
<b>Mol. weight [g/mol]:</b>	378.63
<b>CAS:</b>	305805-39-4

## Physical Properties

Property code	Value	Unit	Source
gf	-18.00	kJ/mol	Joback Method
hf	-667.27	kJ/mol	Joback Method
hfus	63.91	kJ/mol	Joback Method
hvap	84.69	kJ/mol	Joback Method
log10ws	-8.70		Crippen Method
logp	8.133		Crippen Method
mcvol	361.950	ml/mol	McGowan Method
pc	845.05	kPa	Joback Method
rinpol	2798.10		NIST Webbook
rinpol	2798.10		NIST Webbook
tb	883.30	K	Joback Method
tc	1081.46	K	Joback Method
tf	466.29	K	Joback Method
vc	1.427	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1188.91	J/molxK	883.30	Joback Method
cpg	1209.44	J/molxK	916.33	Joback Method
cpg	1228.84	J/molxK	949.35	Joback Method
cpg	1247.18	J/molxK	982.38	Joback Method
cpg	1264.54	J/molxK	1015.41	Joback Method
cpg	1280.98	J/molxK	1048.44	Joback Method
cpg	1296.57	J/molxK	1081.46	Joback Method

dvisc	0.0009593	Paxs	466.29	Joback Method
dvisc	0.0004013	Paxs	535.79	Joback Method
dvisc	0.0002050	Paxs	605.29	Joback Method
dvisc	0.0001203	Paxs	674.79	Joback Method
dvisc	0.0000780	Paxs	744.30	Joback Method
dvisc	0.0000544	Paxs	813.80	Joback Method
dvisc	0.0000402	Paxs	883.30	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C305805394&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C305805394&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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