

# (2R,3R,4R,5R)-3,4-Dimethyl-2,5-bis(3,4,5-trimethoxy)

<b>Inchi:</b>	InChI=1S/C24H32O7/c1-13-14(2)22(16-11-19(27-5)24(30-8)20(12-16)28-6)31-21(13)15-
<b>InchiKey:</b>	ZPINJJOPURFFNV-WJWAULOUA-N
<b>Formula:</b>	C24H32O7
<b>SMILES:</b>	COc1cc(C2OC(c3cc(OC)c(OC)c(OC)c3)C(C)C2C)cc(OC)c1OC
<b>Mol. weight [g/mol]:</b>	432.51
<b>CAS:</b>	50393-99-2

## Physical Properties

Property code	Value	Unit	Source
gf	-384.46	kJ/mol	Joback Method
hf	-1060.31	kJ/mol	Joback Method
hfus	55.92	kJ/mol	Joback Method
hvap	95.84	kJ/mol	Joback Method
log10ws	-5.70		Crippen Method
logp	4.823		Crippen Method
mcvol	331.730	ml/mol	McGowan Method
pc	1126.08	kPa	Joback Method
rinpol	3131.70		NIST Webbook
rinpol	3131.70		NIST Webbook
tb	994.50	K	Joback Method
tc	1223.51	K	Joback Method
tf	646.33	K	Joback Method
vc	1.230	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1133.93	J/molxK	994.50	Joback Method
cpg	1147.28	J/molxK	1032.67	Joback Method
cpg	1158.12	J/molxK	1070.84	Joback Method
cpg	1166.37	J/molxK	1109.01	Joback Method
cpg	1171.96	J/molxK	1147.17	Joback Method
cpg	1174.84	J/molxK	1185.34	Joback Method
cpg	1174.92	J/molxK	1223.51	Joback Method

dvisc	0.0001419	Paxs	646.33	Joback Method
dvisc	0.0001038	Paxs	704.36	Joback Method
dvisc	0.0000796	Paxs	762.39	Joback Method
dvisc	0.0000634	Paxs	820.41	Joback Method
dvisc	0.0000520	Paxs	878.44	Joback Method
dvisc	0.0000437	Paxs	936.47	Joback Method
dvisc	0.0000375	Paxs	994.50	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C50393992&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C50393992&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/95-520-9/2R-3R-4R-5R-3-4-Dimethyl-2-5-bis-3-4-5-trimethoxyphenyl-tetrahydrofuran.p>

Generated by Cheméo on 2024-05-09 17:35:01.03401138 +0000 UTC m=+17565349.954588712.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.