

# Sebacic acid, heptyl 4-iodobenzyl ester

<b>Inchi:</b>	InChI=1S/C24H37IO4/c1-2-3-4-9-12-19-28-23(26)13-10-7-5-6-8-11-14-24(27)29-20-21-1
<b>InchiKey:</b>	NUAMLGYTJKAZAB-UHFFFAOYSA-N
<b>Formula:</b>	C24H37IO4
<b>SMILES:</b>	CCCCCCCOC(=O)CCCCCCCC(=O)OCc1ccc(I)cc1
<b>Mol. weight [g/mol]:</b>	516.45

## Physical Properties

Property code	Value	Unit	Source
gf	-155.74	kJ/mol	Joback Method
hf	-726.36	kJ/mol	Joback Method
hfus	61.55	kJ/mol	Joback Method
hvap	99.64	kJ/mol	Joback Method
log10ws	-8.34		Crippen Method
logp	6.969		Crippen Method
mvol	365.960	ml/mol	McGowan Method
pc	1016.83	kPa	Joback Method
rinpol	3300.00		NIST Webbook
rinpol	3300.00		NIST Webbook
tb	1025.90	K	Joback Method
tc	1256.07	K	Joback Method
tf	601.56	K	Joback Method
vc	1.407	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1170.90	J/molxK	1025.90	Joback Method
cpg	1231.22	J/molxK	1217.71	Joback Method
cpg	1221.67	J/molxK	1179.34	Joback Method
cpg	1210.92	J/molxK	1140.98	Joback Method
cpg	1198.92	J/molxK	1102.62	Joback Method
cpg	1185.60	J/molxK	1064.26	Joback Method
cpg	1239.64	J/molxK	1256.07	Joback Method
dvisc	0.0000207	Paxs	1025.90	Joback Method

dvisc	0.0000269	Paxs	955.18	Joback Method
dvisc	0.0000365	Paxs	884.45	Joback Method
dvisc	0.0000522	Paxs	813.73	Joback Method
dvisc	0.0000800	Paxs	743.01	Joback Method
dvisc	0.0001340	Paxs	672.28	Joback Method
dvisc	0.0002534	Paxs	601.56	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U380629&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U380629&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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