

# Sebacic acid, 4-iodobenzyl 2-methylhex-3-yl ester

Inchi:	InChI=1S/C24H37IO4/c1-4-11-22(19(2)3)29-24(27)13-10-8-6-5-7-9-12-23(26)28-18-20-1
InchiKey:	ZLMNRJOYKQPJNA-UHFFFAOYSA-N
Formula:	C24H37IO4
SMILES:	CCCC(OC(=O)CCCCCCCC(=O)OCc1ccc(I)cc1)C(C)C
Mol. weight [g/mol]:	516.45

## Physical Properties

Property code	Value	Unit	Source
gf	-160.62	kJ/mol	Joback Method
hf	-736.92	kJ/mol	Joback Method
hfus	54.50	kJ/mol	Joback Method
hvap	98.86	kJ/mol	Joback Method
log10ws	-8.21		Crippen Method
logp	6.823		Crippen Method
mvol	365.960	ml/mol	McGowan Method
pc	1027.28	kPa	Joback Method
rinpol	3144.00		NIST Webbook
rinpol	3144.00		NIST Webbook
tb	1025.02	K	Joback Method
tc	1255.64	K	Joback Method
tf	571.56	K	Joback Method
vc	1.395	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1171.51	J/molxK	1025.02	Joback Method
cpg	1230.86	J/molxK	1217.20	Joback Method
cpg	1221.57	J/molxK	1178.76	Joback Method
cpg	1211.06	J/molxK	1140.33	Joback Method
cpg	1199.26	J/molxK	1101.89	Joback Method
cpg	1186.09	J/molxK	1063.46	Joback Method
cpg	1238.98	J/molxK	1255.64	Joback Method
dvisc	0.0000171	Paxs	1025.02	Joback Method

dvisc	0.0000228	Paxs	949.44	Joback Method
dvisc	0.0000320	Paxs	873.87	Joback Method
dvisc	0.0000479	Paxs	798.29	Joback Method
dvisc	0.0000779	Paxs	722.71	Joback Method
dvisc	0.0001422	Paxs	647.14	Joback Method
dvisc	0.0003040	Paxs	571.56	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U380627&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U380627&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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