

# Succinic acid, naphth-2-ylmethyl 2-propylphenyl ester

<b>Inchi:</b>	InChI=1S/C24H24O4/c1-2-7-20-9-5-6-11-22(20)28-24(26)15-14-23(25)27-17-18-12-13-1
<b>InchiKey:</b>	QLSNDJFUDPPUAU-UHFFFAOYSA-N
<b>Formula:</b>	C24H24O4
<b>SMILES:</b>	CCCc1ccccc1OC(=O)CCC(=O)OCc1ccc2ccccc2c1
<b>Mol. weight [g/mol]:</b>	376.44

## Physical Properties

Property code	Value	Unit	Source
gf	-4.43	kJ/mol	Joback Method
hf	-387.10	kJ/mol	Joback Method
hfus	47.81	kJ/mol	Joback Method
hvap	94.85	kJ/mol	Joback Method
log10ws	-7.04		Crippen Method
logp	5.221		Crippen Method
mvol	296.920	ml/mol	McGowan Method
pc	1534.26	kPa	Joback Method
rinpol	3067.00		NIST Webbook
rinpol	3067.00		NIST Webbook
tb	983.40	K	Joback Method
tc	1219.68	K	Joback Method
tf	615.14	K	Joback Method
vc	1.133	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	932.99	J/molxK	983.40	Joback Method
cpg	987.02	J/molxK	1180.30	Joback Method
cpg	978.26	J/molxK	1140.92	Joback Method
cpg	968.57	J/molxK	1101.54	Joback Method
cpg	957.85	J/molxK	1062.16	Joback Method
cpg	946.02	J/molxK	1022.78	Joback Method
cpg	994.92	J/molxK	1219.68	Joback Method
dvisc	0.0000680	Paxs	983.40	Joback Method

dvisc	0.0000832	Paxs	922.02	Joback Method
dvisc	0.0001047	Paxs	860.65	Joback Method
dvisc	0.0001366	Paxs	799.27	Joback Method
dvisc	0.0001861	Paxs	737.89	Joback Method
dvisc	0.0002682	Paxs	676.52	Joback Method
dvisc	0.0004158	Paxs	615.14	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390400&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390400&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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