

# I-Norvaline, N-ethoxycarbonyl-, tetradecyl, ester

Inchi:	InChI=1S/C22H43NO4/c1-4-7-8-9-10-11-12-13-14-15-16-17-19-27-21(24)20(18-5-2)23-2
InchiKey:	OKWHNEMGFDLEJH-UHFFFAOYSA-N
Formula:	C22H43NO4
SMILES:	CCCCCCCCCCCOCC(=O)C(CCC)N=C(O)OCC
Mol. weight [g/mol]:	385.58

## Physical Properties

Property code	Value	Unit	Source
hf	-959.51	kJ/mol	Joback Method
hvap	95.82	kJ/mol	Joback Method
log10ws	-6.57		Crippen Method
logp	6.350		Crippen Method
mcvol	345.700	ml/mol	McGowan Method
pc	917.72	kPa	Joback Method
rinpol	2340.00		NIST Webbook
rinpol	2340.00		NIST Webbook
tb	969.77	K	Joback Method
tc	1192.79	K	Joback Method

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U320709&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U320709&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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