

# Succinic acid, dodec-2-en-1-yl 3-chlorophenyl ester

<b>Inchi:</b>	InChI=1S/C22H31ClO4/c1-2-3-4-5-6-7-8-9-10-11-17-26-21(24)15-16-22(25)27-20-14-12
<b>InchiKey:</b>	SPGFJKZEHMWBTJ-ZHACJKMWSA-N
<b>Formula:</b>	C22H31ClO4
<b>SMILES:</b>	CCCCCCCCC=CCOC(=O)CCC(=O)Oc1cccc(Cl)c1
<b>Mol. weight [g/mol]:</b>	394.93

## Physical Properties

Property code	Value	Unit	Source
gf	-162.41	kJ/mol	Joback Method
hf	-660.47	kJ/mol	Joback Method
hfus	56.36	kJ/mol	Joback Method
hvap	90.16	kJ/mol	Joback Method
log10ws	-7.05		Crippen Method
logp	6.266		Crippen Method
mcvol	319.900	ml/mol	McGowan Method
pc	1183.34	kPa	Joback Method
rinpol	2896.00		NIST Webbook
rinpol	2896.00		NIST Webbook
tb	928.59	K	Joback Method
tc	1140.81	K	Joback Method
tf	545.80	K	Joback Method
vc	1.236	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	996.34	J/molxK	928.59	Joback Method
cpg	1011.37	J/molxK	963.96	Joback Method
cpg	1025.24	J/molxK	999.33	Joback Method
cpg	1038.00	J/molxK	1034.70	Joback Method
cpg	1049.70	J/molxK	1070.07	Joback Method
cpg	1060.39	J/molxK	1105.44	Joback Method
cpg	1070.12	J/molxK	1140.81	Joback Method
dvisc	0.0003584	Paxs	545.80	Joback Method

dvisc	0.0001922	Paxs	609.60	Joback Method
dvisc	0.0001160	Paxs	673.40	Joback Method
dvisc	0.0000764	Paxs	737.19	Joback Method
dvisc	0.0000538	Paxs	800.99	Joback Method
dvisc	0.0000399	Paxs	864.79	Joback Method
dvisc	0.0000308	Paxs	928.59	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389859&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389859&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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