

# 1,2-Cyclohexanedicarboxylic acid, heptyl 3-methylphenyl ester

<b>Inchi:</b>	InChI=1S/C22H32O4/c1-3-4-5-6-9-15-25-21(23)19-13-7-8-14-20(19)22(24)26-18-12-10-
<b>InchiKey:</b>	HZSSQPISMQKMBH-UHFFFAOYSA-N
<b>Formula:</b>	C22H32O4
<b>SMILES:</b>	CCCCCCCOC(=O)C1CCCCC1C(=O)Oc1ccc(C)c1
<b>Mol. weight [g/mol]:</b>	360.49

## Physical Properties

Property code	Value	Unit	Source
gf	-213.96	kJ/mol	Joback Method
hf	-727.97	kJ/mol	Joback Method
hfus	44.87	kJ/mol	Joback Method
hvap	85.94	kJ/mol	Joback Method
log10ws	-5.98		Crippen Method
logp	5.220		Crippen Method
mvol	301.100	ml/mol	McGowan Method
pc	1322.31	kPa	Joback Method
rinpol	2623.00		NIST Webbook
rinpol	2623.00		NIST Webbook
tb	901.88	K	Joback Method
tc	1119.41	K	Joback Method
tf	524.10	K	Joback Method
vc	1.139	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	996.81	J/molxK	901.88	Joback Method
cpg	1013.95	J/molxK	938.13	Joback Method
cpg	1029.50	J/molxK	974.39	Joback Method
cpg	1043.48	J/molxK	1010.64	Joback Method
cpg	1055.92	J/molxK	1046.90	Joback Method
cpg	1066.85	J/molxK	1083.15	Joback Method
cpg	1076.30	J/molxK	1119.41	Joback Method
dvisc	0.0006041	Paxs	524.10	Joback Method

dvisc	0.0003285	Paxs	587.06	Joback Method
dvisc	0.0002010	Paxs	650.03	Joback Method
dvisc	0.0001341	Paxs	712.99	Joback Method
dvisc	0.0000956	Paxs	775.95	Joback Method
dvisc	0.0000717	Paxs	838.92	Joback Method
dvisc	0.0000559	Paxs	901.88	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339834&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339834&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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