

# Isophthalic acid, hexyl oct-3-en-2-yl ester

<b>Inchi:</b>	InChI=1S/C22H32O4/c1-4-6-8-10-13-18(3)26-22(24)20-15-12-14-19(17-20)21(23)25-16-
<b>InchiKey:</b>	RXNBCKGRYOROAL-JLHYAGUSA-N
<b>Formula:</b>	C22H32O4
<b>SMILES:</b>	CCCCC=CC(C)OC(=O)c1cccc(C(=O)OCCCCC)c1
<b>Mol. weight [g/mol]:</b>	360.49

## Physical Properties

Property code	Value	Unit	Source
gf	-152.92	kJ/mol	Joback Method
hf	-650.01	kJ/mol	Joback Method
hfus	48.64	kJ/mol	Joback Method
hvap	85.39	kJ/mol	Joback Method
log10ws	-6.95		Crippen Method
logp	5.715		Crippen Method
mcvol	307.660	ml/mol	McGowan Method
pc	1220.85	kPa	Joback Method
rinpol	2633.00		NIST Webbook
rinpol	2633.00		NIST Webbook
tb	890.72	K	Joback Method
tc	1098.22	K	Joback Method
tf	500.88	K	Joback Method
vc	1.181	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	970.52	J/molxK	890.72	Joback Method
cpg	986.72	J/molxK	925.30	Joback Method
cpg	1001.75	J/molxK	959.89	Joback Method
cpg	1015.64	J/molxK	994.47	Joback Method
cpg	1028.44	J/molxK	1029.05	Joback Method
cpg	1040.19	J/molxK	1063.63	Joback Method
cpg	1050.94	J/molxK	1098.22	Joback Method
dvisc	0.0005015	Paxs	500.88	Joback Method

dvisc	0.0002455	Paxs	565.85	Joback Method
dvisc	0.0001393	Paxs	630.83	Joback Method
dvisc	0.0000878	Paxs	695.80	Joback Method
dvisc	0.0000599	Paxs	760.77	Joback Method
dvisc	0.0000434	Paxs	825.75	Joback Method
dvisc	0.0000330	Paxs	890.72	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U343895&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U343895&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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