

# Isophthalic acid, propyl undec-2-en-1-yl ester

**Inchi:** InChI=1S/C22H32O4/c1-3-5-6-7-8-9-10-11-12-17-26-22(24)20-15-13-14-19(18-20)21(23)  
**InchiKey:** FKGGQZRBLJIGPW-VAWYXSNFSA-N  
**Formula:** C22H32O4  
**SMILES:** CCCCCCCC=CCOC(=O)c1cccc(C(=O)OCCC)c1  
**Mol. weight [g/mol]:** 360.49

## Physical Properties

Property code	Value	Unit	Source
gf	-150.48	kJ/mol	Joback Method
hf	-644.73	kJ/mol	Joback Method
hfus	52.16	kJ/mol	Joback Method
hvap	85.77	kJ/mol	Joback Method
log10ws	-6.83		Crippen Method
logp	5.717		Crippen Method
mcvol	307.660	ml/mol	McGowan Method
pc	1214.05	kPa	Joback Method
rinpol	2747.00		NIST Webbook
rinpol	2747.00		NIST Webbook
tb	891.16	K	Joback Method
tc	1097.33	K	Joback Method
tf	515.88	K	Joback Method
vc	1.188	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	970.03	J/molxK	891.16	Joback Method
cpg	986.16	J/molxK	925.52	Joback Method
cpg	1001.14	J/molxK	959.88	Joback Method
cpg	1015.00	J/molxK	994.25	Joback Method
cpg	1027.80	J/molxK	1028.61	Joback Method
cpg	1039.58	J/molxK	1062.97	Joback Method
cpg	1050.37	J/molxK	1097.33	Joback Method
dvisc	0.0004447	Paxs	515.88	Joback Method

dvisc	0.0002333	Paxs	578.43	Joback Method
dvisc	0.0001388	Paxs	640.97	Joback Method
dvisc	0.0000906	Paxs	703.52	Joback Method
dvisc	0.0000634	Paxs	766.07	Joback Method
dvisc	0.0000468	Paxs	828.61	Joback Method
dvisc	0.0000361	Paxs	891.16	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U343903&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U343903&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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