

# didrovaltrate

<b>Other names:</b>	Dihydroisovaltrate
<b>Inchi:</b>	InChI=1S/C22H32O8/c1-12(2)6-18(24)26-9-15-10-27-21(30-19(25)7-13(3)4)20-16(15)8-
<b>InchiKey:</b>	PHHROXLDZHUIGO-UHFFFAOYSA-N
<b>Formula:</b>	C22H32O8
<b>SMILES:</b>	CC(=O)OC1CC2C(COC(=O)CC(C)C)=COC(OC(=O)CC(C)C)C2C12CO2
<b>Mol. weight [g/mol]:</b>	424.48
<b>CAS:</b>	18296-45-2

## Physical Properties

Property code	Value	Unit	Source
gf	-587.05	kJ/mol	Joback Method
hf	-1279.42	kJ/mol	Joback Method
hfus	56.89	kJ/mol	Joback Method
hvap	99.55	kJ/mol	Joback Method
log10ws	-3.69		Crippen Method
logp	2.742		Crippen Method
mcvol	318.020	ml/mol	McGowan Method
pc	1318.48	kPa	Joback Method
rinpol	2545.70		NIST Webbook
rinpol	2545.70		NIST Webbook
tb	1008.45	K	Joback Method
tc	1237.43	K	Joback Method
tf	652.80	K	Joback Method
vc	1.206	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1150.61	J/molxK	1008.45	Joback Method
cpg	1170.80	J/molxK	1046.61	Joback Method
cpg	1190.70	J/molxK	1084.78	Joback Method
cpg	1210.45	J/molxK	1122.94	Joback Method
cpg	1230.24	J/molxK	1161.10	Joback Method
cpg	1250.23	J/molxK	1199.26	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18296452&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18296452&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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