

# Glutaric acid, 2-ethylhexyl 3-phenylpropyl ester

Inchi:	InChI=1S/C22H34O4/c1-3-5-11-19(4-2)18-26-22(24)16-9-15-21(23)25-17-10-14-20-12-7
InchiKey:	SVBYHUSZHNULDV-UHFFFAOYSA-N
Formula:	C22H34O4
SMILES:	CCCCC(CC)COC(=O)CCCC(=O)OCCc1ccccc1
Mol. weight [g/mol]:	362.50

## Physical Properties

Property code	Value	Unit	Source
gf	-223.51	kJ/mol	Joback Method
hf	-755.76	kJ/mol	Joback Method
hfus	48.83	kJ/mol	Joback Method
hvap	84.77	kJ/mol	Joback Method
log10ws	-5.62		Crippen Method
logp	5.092		Crippen Method
mvol	311.960	ml/mol	McGowan Method
pc	1189.88	kPa	Joback Method
rinpol	2607.00		NIST Webbook
rinpol	2607.00		NIST Webbook
tb	881.58	K	Joback Method
tc	1084.93	K	Joback Method
tf	493.44	K	Joback Method
vc	1.202	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	998.33	J/molxK	881.58	Joback Method
cpg	1069.80	J/molxK	1051.04	Joback Method
cpg	1057.86	J/molxK	1017.15	Joback Method
cpg	1044.77	J/molxK	983.26	Joback Method
cpg	1030.51	J/molxK	949.36	Joback Method
cpg	1015.04	J/molxK	915.47	Joback Method
cpg	1080.63	J/molxK	1084.93	Joback Method
dvisc	0.0000371	Paxs	881.58	Joback Method

dvisc	0.0000495	Paxs	816.89	Joback Method
dvisc	0.0000692	Paxs	752.20	Joback Method
dvisc	0.0001031	Paxs	687.51	Joback Method
dvisc	0.0001670	Paxs	622.82	Joback Method
dvisc	0.0003024	Paxs	558.13	Joback Method
dvisc	0.0006396	Paxs	493.44	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391776&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391776&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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