

# Pimelic acid, 4-methyl-2-pentyl 3-phenylpropyl ester

<b>Inchi:</b>	InChI=1S/C22H34O4/c1-18(2)17-19(3)26-22(24)15-9-5-8-14-21(23)25-16-10-13-20-11-6
<b>InchiKey:</b>	GXAGWHDYQDXRMW-UHFFFAOYSA-N
<b>Formula:</b>	C22H34O4
<b>SMILES:</b>	CC(C)CC(C)OC(=O)CCCCC(=O)OCCc1ccccc1
<b>Mol. weight [g/mol]:</b>	362.50

## Physical Properties

Property code	Value	Unit	Source
gf	-225.95	kJ/mol	Joback Method
hf	-761.04	kJ/mol	Joback Method
hfus	45.30	kJ/mol	Joback Method
hvap	84.38	kJ/mol	Joback Method
log10ws	-5.73		Crippen Method
logp	5.091		Crippen Method
mvol	311.960	ml/mol	McGowan Method
pc	1196.48	kPa	Joback Method
rinpol	2584.00		NIST Webbook
rinpol	2584.00		NIST Webbook
tb	881.14	K	Joback Method
tc	1085.73	K	Joback Method
tf	478.44	K	Joback Method
vc	1.196	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	998.81	J/molxK	881.14	Joback Method
cpg	1015.61	J/molxK	915.24	Joback Method
cpg	1031.13	J/molxK	949.34	Joback Method
cpg	1045.41	J/molxK	983.44	Joback Method
cpg	1058.50	J/molxK	1017.53	Joback Method
cpg	1070.42	J/molxK	1051.63	Joback Method
cpg	1081.21	J/molxK	1085.73	Joback Method
dvisc	0.0007415	Paxs	478.44	Joback Method

dvisc	0.0003233	Paxs	545.56	Joback Method
dvisc	0.0001691	Paxs	612.67	Joback Method
dvisc	0.0001005	Paxs	679.79	Joback Method
dvisc	0.0000656	Paxs	746.91	Joback Method
dvisc	0.0000459	Paxs	814.02	Joback Method
dvisc	0.0000340	Paxs	881.14	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U416511&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U416511&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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