

# Diglycolic acid, decyl 2,5-dimethylphenyl ester

<b>Inchi:</b>	InChI=1S/C22H34O5/c1-4-5-6-7-8-9-10-11-14-26-21(23)16-25-17-22(24)27-20-15-18(2)
<b>InchiKey:</b>	YBBXWOSIJWOQMD-UHFFFAOYSA-N
<b>Formula:</b>	C22H34O5
<b>SMILES:</b>	CCCCCCCCCOC(=O)COCC(=O)Oc1cc(C)ccc1C
<b>Mol. weight [g/mol]:</b>	378.50

## Physical Properties

Property code	Value	Unit	Source
gf	-345.33	kJ/mol	Joback Method
hf	-905.64	kJ/mol	Joback Method
hfus	52.76	kJ/mol	Joback Method
hvap	88.89	kJ/mol	Joback Method
log10ws	-5.71		Crippen Method
logp	4.909		Crippen Method
mcvol	317.830	ml/mol	McGowan Method
pc	1145.99	kPa	Joback Method
rinsol	3317.00		NIST Webbook
tb	914.40	K	Joback Method
tc	1121.51	K	Joback Method
tf	555.71	K	Joback Method
vc	1.226	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1026.34	J/molxK	914.40	Joback Method
cpg	1092.25	J/molxK	1086.99	Joback Method
cpg	1081.77	J/molxK	1052.47	Joback Method
cpg	1069.94	J/molxK	1017.95	Joback Method
cpg	1056.77	J/molxK	983.44	Joback Method
cpg	1042.24	J/molxK	948.92	Joback Method
cpg	1101.41	J/molxK	1121.51	Joback Method
dvisc	0.0000313	Paxs	914.40	Joback Method
dvisc	0.0000399	Paxs	854.62	Joback Method

dvisc	0.0000528	Paxs	794.84	Joback Method
dvisc	0.0000730	Paxs	735.05	Joback Method
dvisc	0.0001070	Paxs	675.27	Joback Method
dvisc	0.0001688	Paxs	615.49	Joback Method
dvisc	0.0002938	Paxs	555.71	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382712&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382712&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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