

# Succinic acid, 3-methylbut-2-yl diphenylmethyl ester

<b>Inchi:</b>	InChI=1S/C22H26O4/c1-16(2)17(3)25-20(23)14-15-21(24)26-22(18-10-6-4-7-11-18)19-1
<b>InchiKey:</b>	NKWSYGOWFQPFDT-UHFFFAOYSA-N
<b>Formula:</b>	C22H26O4
<b>SMILES:</b>	CC(C)C(C)OC(=O)CCC(=O)OC(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	354.44

## Physical Properties

Property code	Value	Unit	Source
gf	-115.98	kJ/mol	Joback Method
hf	-529.79	kJ/mol	Joback Method
hfus	35.82	kJ/mol	Joback Method
hvap	86.27	kJ/mol	Joback Method
log10ws	-5.39		Crippen Method
logp	4.687		Crippen Method
mcvol	288.200	ml/mol	McGowan Method
pc	1535.46	kPa	Joback Method
rinpol	2518.00		NIST Webbook
rinpol	2518.00		NIST Webbook
tb	907.38	K	Joback Method
tc	1135.48	K	Joback Method
tf	489.86	K	Joback Method
vc	1.081	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	900.38	J/molxK	907.38	Joback Method
cpg	915.18	J/molxK	945.40	Joback Method
cpg	928.54	J/molxK	983.41	Joback Method
cpg	940.52	J/molxK	1021.43	Joback Method
cpg	951.18	J/molxK	1059.45	Joback Method
cpg	960.58	J/molxK	1097.46	Joback Method
cpg	968.77	J/molxK	1135.48	Joback Method
dvisc	0.0007036	Paxs	489.86	Joback Method

dvisc	0.0003045	Paxs	559.45	Joback Method
dvisc	0.0001586	Paxs	629.03	Joback Method
dvisc	0.0000941	Paxs	698.62	Joback Method
dvisc	0.0000613	Paxs	768.21	Joback Method
dvisc	0.0000429	Paxs	837.79	Joback Method
dvisc	0.0000317	Paxs	907.38	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390165&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390165&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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