

Terephthalic acid, 4-nitro-3-methylbenzyl pentyl ester

Inchi:	InChI=1S/C21H23NO6/c1-3-4-5-12-27-20(23)17-7-9-18(10-8-17)21(24)28-14-16-6-11-19
InchiKey:	FXNNYSBTYNAFCT-UHFFFAOYSA-N
Formula:	C21H23NO6
SMILES:	CCCCCOC(=O)c1ccc(C(=O)OCc2ccc([N+](=O)[O-])c(C)c2)cc1
Mol. weight [g/mol]:	385.41

Physical Properties

Property code	Value	Unit	Source
gf	-110.42	kJ/mol	Joback Method
hf	-538.48	kJ/mol	Joback Method
hfus	54.00	kJ/mol	Joback Method
hvap	103.78	kJ/mol	Joback Method
log10ws	-6.87		Crippen Method
logp	4.607		Crippen Method
mvol	291.530	ml/mol	McGowan Method
pc	1583.49	kPa	Joback Method
rinpol	3299.00		NIST Webbook
rinpol	3299.00		NIST Webbook
tb	1052.60	K	Joback Method
tc	1298.25	K	Joback Method
tf	704.76	K	Joback Method
vc	1.125	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	932.47	J/molxK	1052.60	Joback Method
cpg	942.10	J/molxK	1093.54	Joback Method
cpg	950.22	J/molxK	1134.48	Joback Method
cpg	956.89	J/molxK	1175.42	Joback Method
cpg	962.15	J/molxK	1216.37	Joback Method
cpg	966.03	J/molxK	1257.31	Joback Method
cpg	968.60	J/molxK	1298.25	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U416102&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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