

# cis-Cyclohex-4-en-1,2-dicarboxylic acid, 4-chloro-3-methylphenyl hexyl ester

<b>Inchi:</b>	InChI=1S/C21H27ClO4/c1-3-4-5-8-13-25-20(23)17-9-6-7-10-18(17)21(24)26-16-11-12-19
<b>InchiKey:</b>	UFKCGOZAOHSYHY-UHFFFAOYSA-N
<b>Formula:</b>	C21H27ClO4
<b>SMILES:</b>	CCCCCOC(=O)C1CC=CCC1C(=O)Oc1ccc(Cl)c(C)c1
<b>Mol. weight [g/mol]:</b>	378.89

## Physical Properties

Property code	Value	Unit	Source
gf	-213.98	kJ/mol	Joback Method
hf	-676.76	kJ/mol	Joback Method
hfus	47.31	kJ/mol	Joback Method
hvap	89.05	kJ/mol	Joback Method
log10ws	-6.10		Crippen Method
logp	5.260		Crippen Method
mvol	294.950	ml/mol	McGowan Method
pc	1402.74	kPa	Joback Method
rinpol	2707.00		NIST Webbook
rinpol	2707.00		NIST Webbook
tb	920.57	K	Joback Method
tc	1144.42	K	Joback Method
tf	556.03	K	Joback Method
vc	1.119	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	932.20	J/molxK	920.57	Joback Method
cpg	947.01	J/molxK	957.88	Joback Method
cpg	960.28	J/molxK	995.19	Joback Method
cpg	972.02	J/molxK	1032.50	Joback Method
cpg	982.26	J/molxK	1069.81	Joback Method
cpg	991.03	J/molxK	1107.11	Joback Method
cpg	998.36	J/molxK	1144.42	Joback Method
dvisc	0.0004891	Paxs	556.03	Joback Method

dvisc	0.0002922	Paxs	616.79	Joback Method
dvisc	0.0001915	Paxs	677.54	Joback Method
dvisc	0.0001345	Paxs	738.30	Joback Method
dvisc	0.0000997	Paxs	799.06	Joback Method
dvisc	0.0000771	Paxs	859.81	Joback Method
dvisc	0.0000617	Paxs	920.57	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382651&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382651&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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