

# Octabenzone

**Other names:**

Methanone, [2-hydroxy-4-(octyloxy)phenyl]phenyl-  
Benzophenone, 2-hydroxy-4-(octyloxy)-  
Benzon OO  
Benzophenone-12  
Cyasorb UV 531  
Spectra-Sorb UV 531  
UF 4  
UV 531  
2-Benzoyl-5-(octyloxy)phenol  
2-Hydroxy-4-(n-octoxy)benzophenone  
2-Hydroxy-4-(n-octyloxy)benzophenone  
2-Hydroxy-4-(octoxy)benzophenone  
2-Hydroxy-4-(octyloxy)benzophenone  
4-(n-Octyloxy)-2-hydroxybenzophenone  
4-(Octoxy)-2-hydroxybenzophenone  
4-(Octyloxy)-2-hydroxybenzophenone  
UV 1  
2-Hydroxy-4-oktyloxybenzofenon  
Octabenzon  
[2-Hydroxy-4-(octyloxy)phenyl]phenylmethanone  
Hostavin ARO 8  
Lowilite 22  
Specta-srob UV 531  
Uvinul 408  
Aduvex 248  
Advastab 46  
Anti-UV P  
Carstab 700  
Chimassorb 81  
Mark 1413  
NSC 163400  
Rhodialux P  
Sanduvor 3035  
Seikalizer E  
Sumisorb 130  
UV 1 (ultraviolet absorber)  
Uvinul M 408  
Viosorb 130  
Zislizer E

**Inchi:**

InChI=1S/C21H26O3/c1-2-3-4-5-6-10-15-24-18-13-14-19(20(22)16-18)21(23)17-11-8-7-9

**InchiKey:** QUAMTGJKVDWJEQ-UHFFFAOYSA-N  
**Formula:** C<sub>21</sub>H<sub>26</sub>O<sub>3</sub>  
**SMILES:** CCCCCCOC1CCC(C(=O)C2CCCCC2)C(O)C1  
**Mol. weight [g/mol]:** 326.43  
**CAS:** 1843-05-6

## Physical Properties

Property code	Value	Unit	Source
gf	-47.41	kJ/mol	Joback Method
hf	-437.29	kJ/mol	Joback Method
h <sub>fus</sub>	46.41	kJ/mol	Joback Method
h <sub>vap</sub>	89.72	kJ/mol	Joback Method
log <sub>10</sub> ws	-6.03		Crippen Method
logp	5.363		Crippen Method
m <sub>cvol</sub>	272.540	ml/mol	McGowan Method
pc	1747.74	kPa	Joback Method
tb	895.13	K	Joback Method
tc	1120.71	K	Joback Method
tf	575.67	K	Joback Method
vc	0.986	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
c <sub>pg</sub>	851.21	J/mol×K	895.13	Joback Method
c <sub>pg</sub>	921.54	J/mol×K	1083.11	Joback Method
c <sub>pg</sub>	908.78	J/mol×K	1045.52	Joback Method
c <sub>pg</sub>	895.47	J/mol×K	1007.92	Joback Method
c <sub>pg</sub>	881.51	J/mol×K	970.32	Joback Method
c <sub>pg</sub>	866.79	J/mol×K	932.73	Joback Method
c <sub>pg</sub>	933.86	J/mol×K	1120.71	Joback Method
d <sub>visc</sub>	0.0000029	Paxs	895.13	Joback Method
d <sub>visc</sub>	0.0000042	Paxs	841.89	Joback Method
d <sub>visc</sub>	0.0000063	Paxs	788.64	Joback Method
d <sub>visc</sub>	0.0000102	Paxs	735.40	Joback Method
d <sub>visc</sub>	0.0000177	Paxs	682.16	Joback Method
d <sub>visc</sub>	0.0000336	Paxs	628.91	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1843056&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1843056&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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