

# Pimelic acid, butyl 2-naphthyl ester

<b>Inchi:</b>	InChI=1S/C21H26O4/c1-2-3-15-24-20(22)11-5-4-6-12-21(23)25-19-14-13-17-9-7-8-10-18
<b>InchiKey:</b>	YXROBZJZPVMICO-UHFFFAOYSA-N
<b>Formula:</b>	C21H26O4
<b>SMILES:</b>	CCCCOC(=O)CCCCC(=O)Oc1ccc2ccccc2c1
<b>Mol. weight [g/mol]:</b>	342.43

## Physical Properties

Property code	Value	Unit	Source
gf	-132.47	kJ/mol	Joback Method
hf	-550.24	kJ/mol	Joback Method
hfus	46.39	kJ/mol	Joback Method
hvap	85.23	kJ/mol	Joback Method
log10ws	-6.22		Crippen Method
logp	5.039		Crippen Method
mvol	278.410	ml/mol	McGowan Method
pc	1501.15	kPa	Joback Method
rinpol	2786.00		NIST Webbook
rinpol	2786.00		NIST Webbook
tb	883.10	K	Joback Method
tc	1096.62	K	Joback Method
tf	542.39	K	Joback Method
vc	1.073	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	861.05	J/molxK	883.10	Joback Method
cpg	875.78	J/molxK	918.69	Joback Method
cpg	889.42	J/molxK	954.27	Joback Method
cpg	902.02	J/molxK	989.86	Joback Method
cpg	913.64	J/molxK	1025.45	Joback Method
cpg	924.33	J/molxK	1061.03	Joback Method
cpg	934.15	J/molxK	1096.62	Joback Method
dvisc	0.0006540	Paxs	542.39	Joback Method

dvisc	0.0004074	Paxs	599.17	Joback Method
dvisc	0.0002754	Paxs	655.96	Joback Method
dvisc	0.0001982	Paxs	712.74	Joback Method
dvisc	0.0001497	Paxs	769.53	Joback Method
dvisc	0.0001176	Paxs	826.31	Joback Method
dvisc	0.0000952	Paxs	883.10	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U416709&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U416709&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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