

# Isophthalic acid, (2-chlorocyclohexyl)methyl isoheptyl ester

<b>Inchi:</b>	InChI=1S/C21H29ClO4/c1-15(2)7-6-12-25-20(23)16-9-5-10-17(13-16)21(24)26-14-18-8-3
<b>InchiKey:</b>	ANCPDJAIGGZLQF-UHFFFAOYSA-N
<b>Formula:</b>	C21H29ClO4
<b>SMILES:</b>	CC(C)CCCOC(=O)c1cccc(C(=O)OCC2CCCCC2Cl)c1
<b>Mol. weight [g/mol]:</b>	380.91

## Physical Properties

Property code	Value	Unit	Source
gf	-236.75	kJ/mol	Joback Method
hf	-728.35	kJ/mol	Joback Method
hfus	42.95	kJ/mol	Joback Method
hvap	87.71	kJ/mol	Joback Method
log10ws	-6.24		Crippen Method
logp	5.234		Crippen Method
mvol	299.250	ml/mol	McGowan Method
pc	1389.18	kPa	Joback Method
rinpol	2927.00		NIST Webbook
rinpol	2927.00		NIST Webbook
tb	915.99	K	Joback Method
tc	1140.34	K	Joback Method
tf	527.75	K	Joback Method
vc	1.127	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	963.64	J/molxK	915.99	Joback Method
cpg	1026.44	J/molxK	1102.95	Joback Method
cpg	1017.01	J/molxK	1065.56	Joback Method
cpg	1006.05	J/molxK	1028.17	Joback Method
cpg	993.52	J/molxK	990.77	Joback Method
cpg	979.39	J/molxK	953.38	Joback Method
cpg	1034.38	J/molxK	1140.34	Joback Method
dvisc	0.0000506	Paxs	915.99	Joback Method

dvisc	0.0000655	Paxs	851.28	Joback Method
dvisc	0.0000883	Paxs	786.58	Joback Method
dvisc	0.0001256	Paxs	721.87	Joback Method
dvisc	0.0001914	Paxs	657.16	Joback Method
dvisc	0.0003200	Paxs	592.46	Joback Method
dvisc	0.0006069	Paxs	527.75	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U343796&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U343796&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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