

# 4-Pentenoic acid, 2-methyl-, pentadecyl ester

**Inchi:** InChI=1S/C21H40O2/c1-4-6-7-8-9-10-11-12-13-14-15-16-17-19-23-21(22)20(3)18-5-2/h5  
**InchiKey:** NQUFOCGJLQNROD-UHFFFAOYSA-N  
**Formula:** C21H40O2  
**SMILES:** C=CCC(C)C(=O)OCCCCCCCCCCCCCCC  
**Mol. weight [g/mol]:** 324.54

## Physical Properties

Property code	Value	Unit	Source
gf	-22.58	kJ/mol	Joback Method
hf	-601.42	kJ/mol	Joback Method
hfus	48.13	kJ/mol	Joback Method
hvap	70.44	kJ/mol	Joback Method
log10ws	-7.09		Crippen Method
logp	6.833		Crippen Method
mcvol	309.890	ml/mol	McGowan Method
pc	1015.53	kPa	Joback Method
rinpol	2214.00		NIST Webbook
tb	752.41	K	Joback Method
tc	927.62	K	Joback Method
tf	381.83	K	Joback Method
vc	1.210	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	943.77	J/molxK	752.41	Joback Method
cpg	963.48	J/molxK	781.61	Joback Method
cpg	982.23	J/molxK	810.81	Joback Method
cpg	1000.04	J/molxK	840.01	Joback Method
cpg	1016.94	J/molxK	869.21	Joback Method
cpg	1032.97	J/molxK	898.42	Joback Method
cpg	1048.15	J/molxK	927.62	Joback Method
dvisc	0.0018585	Paxs	381.83	Joback Method
dvisc	0.0007126	Paxs	443.59	Joback Method

dvisc	0.0003454	Paxs	505.36	Joback Method
dvisc	0.0001960	Paxs	567.12	Joback Method
dvisc	0.0001243	Paxs	628.88	Joback Method
dvisc	0.0000856	Paxs	690.65	Joback Method
dvisc	0.0000626	Paxs	752.41	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406118&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406118&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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