

# 4-(4-Chloro-2-methylphenoxy)butyric acid, octyl ester

Inchi:	InChI=1S/C19H29ClO3/c1-3-4-5-6-7-8-13-23-19(21)10-9-14-22-18-12-11-17(20)15-16(18)
InchiKey:	WROBQVBWQOGTTQ-UHFFFAOYSA-N
Formula:	C19H29ClO3
SMILES:	CCCCCCCCOC(=O)CCCOc1ccc(Cl)cc1C
Mol. weight [g/mol]:	340.88

## Physical Properties

Property code	Value	Unit	Source
gf	-148.60	kJ/mol	Joback Method
hf	-614.66	kJ/mol	Joback Method
hfus	46.40	kJ/mol	Joback Method
hvap	77.44	kJ/mol	Joback Method
log10ws	-6.22		Crippen Method
logp	5.711		Crippen Method
mvol	280.360	ml/mol	McGowan Method
pc	1323.28	kPa	Joback Method
rinpol	3000.00		NIST Webbook
rinpol	3000.00		NIST Webbook
tb	806.90	K	Joback Method
tc	1004.33	K	Joback Method
tf	479.66	K	Joback Method
vc	1.083	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	827.81	J/molxK	806.90	Joback Method
cpg	899.36	J/molxK	971.43	Joback Method
cpg	887.07	J/molxK	938.52	Joback Method
cpg	873.79	J/molxK	905.62	Joback Method
cpg	859.50	J/molxK	872.71	Joback Method
cpg	844.18	J/molxK	839.81	Joback Method
cpg	910.67	J/molxK	1004.33	Joback Method
dvisc	0.0000572	Paxs	806.90	Joback Method

dvisc	0.0000729	Paxs	752.36	Joback Method
dvisc	0.0000964	Paxs	697.82	Joback Method
dvisc	0.0001338	Paxs	643.28	Joback Method
dvisc	0.0001971	Paxs	588.74	Joback Method
dvisc	0.0003144	Paxs	534.20	Joback Method
dvisc	0.0005575	Paxs	479.66	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U415084&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U415084&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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