

Fluoranthene, 2-nitro-

| | |
|-----------------------------|--|
| Other names: | 2-Nitrofluoranthene |
| Inchi: | InChI=1S/C16H9NO2/c18-17(19)11-8-10-4-3-7-14-12-5-1-2-6-13(12)15(9-11)16(10)14/h |
| InchiKey: | VBCBFNMZBHKVQN-UHFFFAOYSA-N |
| Formula: | C16H9NO2 |
| SMILES: | O=[N+]([O-])c1cc2c3c(cccc3c1)-c1cccc1-2 |
| Mol. weight [g/mol]: | 247.25 |
| CAS: | 13177-29-2 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | 517.10 | kJ/mol | Joback Method |
| hf | 345.54 | kJ/mol | Joback Method |
| hfus | 35.47 | kJ/mol | Joback Method |
| hvap | 76.35 | kJ/mol | Joback Method |
| log10ws | -7.38 | | Crippen Method |
| logp | 4.395 | | Crippen Method |
| mcvol | 175.880 | ml/mol | McGowan Method |
| pc | 3062.55 | kPa | Joback Method |
| rinpol | 412.45 | | NIST Webbook |
| rinpol | 411.26 | | NIST Webbook |
| rinpol | 412.59 | | NIST Webbook |
| rinpol | 412.48 | | NIST Webbook |
| rinpol | 411.85 | | NIST Webbook |
| rinpol | 411.26 | | NIST Webbook |
| rinpol | 411.85 | | NIST Webbook |
| rinpol | 412.45 | | NIST Webbook |
| tb | 808.18 | K | Joback Method |
| tc | 1082.43 | K | Joback Method |
| tf | 582.05 | K | Joback Method |
| vc | 0.702 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-------|------|-----------------|--------|
|---------------|-------|------|-----------------|--------|

| | | | | |
|-----|--------|---------|---------|---------------|
| cpg | 473.39 | J/mol×K | 808.18 | Joback Method |
| cpg | 484.71 | J/mol×K | 853.89 | Joback Method |
| cpg | 495.53 | J/mol×K | 899.60 | Joback Method |
| cpg | 506.09 | J/mol×K | 945.31 | Joback Method |
| cpg | 516.66 | J/mol×K | 991.01 | Joback Method |
| cpg | 527.49 | J/mol×K | 1036.72 | Joback Method |
| cpg | 538.85 | J/mol×K | 1082.43 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C13177292&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

Latest version available from:

<https://www.chemeo.com/cid/96-408-3/Fluoranthene-2-nitro.pdf>

Generated by Cheméo on 2024-04-26 21:23:32.377546858 +0000 UTC m=+16455861.298124173.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.