

# Carbonic acid, monoamide, N-nonyl-, decyl ester

Inchi:	InChI=1S/C20H41NO2/c1-3-5-7-9-11-13-15-17-19-23-20(22)21-18-16-14-12-10-8-6-4-2/
InchiKey:	ZWBDYIKSAYVLBG-UHFFFAOYSA-N
Formula:	C20H41NO2
SMILES:	CCCCCCCCCOC(O)=NCCCCCCCCC
Mol. weight [g/mol]:	327.55

## Physical Properties

Property code	Value	Unit	Source
hf	-668.15	kJ/mol	Joback Method
hvap	82.60	kJ/mol	Joback Method
log10ws	-6.76		Crippen Method
logp	6.808		Crippen Method
mcvol	310.080	ml/mol	McGowan Method
pc	999.54	kPa	Joback Method
rinsol	2625.00		NIST Webbook
rinsol	2625.00		NIST Webbook
tb	848.16	K	Joback Method
tc	1038.42	K	Joback Method

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U415184&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U415184&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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