

# Nisoldipine

<b>Other names:</b>	3,5-Pyridinedicarboxylic acid, 1,4-dihydro-2,6-dimethyl-4-(2-nitrophenyl)-, methyl 2-methylpropyl ester BAY-k 5552 3,5-Pyridinedicarboxylic acid, 1,4-dihydro-2,6-dimethyl-4-(2-nitrophenyl)-, methyl 2-methylpropyl ester, (.+/-.)- (.+/-.)-Isobutyl methyl 1,4-dihydro-2,6-dimethyl-4-(o-nitrophenyl)-3,5-pyridinedicarboxylate 1,4-Dihydro-2,6-dimethyl-4-(2-nitrophenyl)-3,5-pyridinedicarboxylic acid methyl 2-methylpropyl ester Baymycard Nisocor Sular Syscor Zadipina (.+/-.)-Nisoldipine
<b>Inchi:</b>	InChI=1S/C20H24N2O6/c1-11(2)10-28-20(24)17-13(4)21-12(3)16(19(23)27-5)18(17)14-8
<b>InchiKey:</b>	VKQFCGNPDRICFG-UHFFFAOYSA-N
<b>Formula:</b>	C20H24N2O6
<b>SMILES:</b>	<chem>COC(=O)C1=C(C)NC(C)=C(C(=O)OCC(C)C)C1c1cccc1[N+](=O)[O-]</chem>
<b>Mol. weight [g/mol]:</b>	388.41
<b>CAS:</b>	63675-72-9

## Physical Properties

Property code	Value	Unit	Source
gf	-80.87	kJ/mol	Joback Method
hf	-574.90	kJ/mol	Joback Method
hfus	56.93	kJ/mol	Joback Method
hvap	107.99	kJ/mol	Joback Method
log10ws	-5.18		Crippen Method
logp	3.202		Crippen Method
mcvol	291.720	ml/mol	McGowan Method
pc	1641.76	kPa	Joback Method
rinpol	2679.00		NIST Webbook
tb	1078.98	K	Joback Method
tc	1331.30	K	Joback Method
tf	791.04	K	Joback Method
vc	1.113	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	953.57	J/mol×K	1078.98	Joback Method
cpg	961.69	J/mol×K	1121.03	Joback Method
cpg	967.83	J/mol×K	1163.09	Joback Method
cpg	971.99	J/mol×K	1205.14	Joback Method
cpg	974.15	J/mol×K	1247.19	Joback Method
cpg	974.34	J/mol×K	1289.25	Joback Method
cpg	972.55	J/mol×K	1331.30	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C63675729&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C63675729&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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