L-Leucine, N-butoxycarbonyl-N-methyl-, octyl ester

Inchi: InChI=1S/C20H39NO4/c1-6-8-10-11-12-13-15-24-19(22)18(16-17(3)4)21(5)20(23)25-14-

InchiKey: SLXXTLCHZFCXRO-UHFFFAOYSA-N

Formula: C20H39NO4

SMILES: CCCCCCCCOC(=O)C(CC(C)C)N(C)C(=O)OCCCC

Mol. weight [g/mol]: 357.53

Physical Properties

Property code	Value	Unit	Source
hf	-1031.20	kJ/mol	Cheméo Relay (1.0)
hfus	49.10	kJ/mol	Joback Method
hvap	97.80	kJ/mol	Cheméo Relay (1.0)
log10ws	-4.75		Cheméo Relay (1.0)
logp	5.173		Crippen Method
mcvol	317.520	ml/mol	McGowan Method
rinpol	2189.00		NIST Webbook
rinpol	2189.00		NIST Webbook
tb	610.77	K	Cheméo Relay (1.0)
tf	249.33	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1013.53	J/mol×K	821.14	Joback Method
cpg	1032.07	J/mol×K	852.25	Joback Method
cpg	1049.49	J/mol×K	883.35	Joback Method
cpg	1065.82	J/mol×K	914.46	Joback Method
cpg	1081.08	J/mol×K	945.57	Joback Method
cpg	1095.30	J/mol×K	976.68	Joback Method
cpg	1108.50	J/mol×K	1007.78	Joback Method

Sources

Cheméo Relay (1.0, beta):	
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U321887&Units=SI
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0):	

Legend

cpg:	Ideal gas heat capacity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tf:	Normal melting (fusion) point