

1-iodo-2-trifluoromethylpropane

Other names:	1,1,1-Trifluoro-3-iodo-2-methylpropane
Inchi:	InChI=1S/C4H6F3I/c1-3(2-8)4(5,6)7/h3H,2H2,1H3
InchiKey:	UEEFSGKHKKFYRL-UHFFFAOYSA-N
Formula:	C4H6F3I
SMILES:	CC(Cl)C(F)(F)F
Mol. weight [g/mol]:	237.99
CAS:	26653-47-4

Physical Properties

Property code	Value	Unit	Source
hfus	8.82	kJ/mol	Joback Method
hvap	34.83	kJ/mol	Cheméo Relay (1.0)
ie	9.51	eV	Cheméo Relay (1.0)
logp	2.620		Crippen Method
mcvol	98.350	ml/mol	McGowan Method
tb	383.80	K	Cheméo Relay (1.0)
tf	177.87	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	163.64	J/mol×K	378.20	Joback Method
cpg	172.21	J/mol×K	409.59	Joback Method
cpg	180.19	J/mol×K	440.98	Joback Method
cpg	187.63	J/mol×K	472.36	Joback Method
cpg	194.53	J/mol×K	503.75	Joback Method
cpg	200.94	J/mol×K	535.14	Joback Method
cpg	206.89	J/mol×K	566.53	Joback Method
hvapt	30.40	kJ/mol	333.00	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.45084e+01
Coeff. B	-3.65190e+03
Temperature range (K), min.	256.80
Temperature range (K), max.	397.08

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0):	
Cheméo Relay (1.0, beta):	
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C26653474&Units=SI
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tf:	Normal melting (fusion) point

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