

# Propanoic acid, 3-bromo-2-chloro, ethyl ester

<b>Inchi:</b>	InChI=1S/C5H8BrClO2/c1-2-9-5(8)4(7)3-6/h4H,2-3H2,1H3
<b>InchiKey:</b>	QHNYTHIGYIAQCD-UHFFFAOYSA-N
<b>Formula:</b>	C5H8BrClO2
<b>SMILES:</b>	CCOC(=O)C(Cl)CBr
<b>Mol. weight [g/mol]:</b>	215.47

## Physical Properties

Property code	Value	Unit	Source
gf	-242.75	kJ/mol	Joback Method
hf	-386.02	kJ/mol	Joback Method
hfus	17.45	kJ/mol	Joback Method
hvap	46.31	kJ/mol	Joback Method
log10ws	-1.47		Crippen Method
logp	1.552		Crippen Method
mcvol	118.490	ml/mol	McGowan Method
pc	3848.31	kPa	Joback Method
tb	493.24	K	Joback Method
tc	700.45	K	Joback Method
tf	292.99	K	Joback Method
vc	0.445	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	218.20	J/molxK	493.24	Joback Method
cpg	226.55	J/molxK	527.77	Joback Method
cpg	234.49	J/molxK	562.31	Joback Method
cpg	242.03	J/molxK	596.84	Joback Method
cpg	249.18	J/molxK	631.38	Joback Method
cpg	255.93	J/molxK	665.91	Joback Method
cpg	262.31	J/molxK	700.45	Joback Method
dvisc	0.0032213	Paxs	292.99	Joback Method
dvisc	0.0018152	Paxs	326.37	Joback Method
dvisc	0.0011377	Paxs	359.74	Joback Method

dvisc	0.0007719	Paxs	393.12	Joback Method
dvisc	0.0005566	Paxs	426.49	Joback Method
dvisc	0.0004208	Paxs	459.87	Joback Method
dvisc	0.0003304	Paxs	493.24	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R30296&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R30296&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/96-736-9/Propanoic-acid-3-bromo-2-chloro-ethyl-ester.pdf>

Generated by Cheméo on 2024-04-27 17:50:41.710757413 +0000 UTC m=+16529490.631334729.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.