

# 2-Propanol, 1-chloro-3-ethoxy-

<b>Other names:</b>	1-Ethoxy-2-hydroxy-3-chloropropane 3-Ethoxy-1-chloro-2-propanol 1-Chloro-3-ethoxy-2-propanol
<b>Inchi:</b>	InChI=1S/C5H11ClO2/c1-2-8-4-5(7)3-6/h5,7H,2-4H2,1H3
<b>InchiKey:</b>	XHIINWKFCZSGNY-UHFFFAOYSA-N
<b>Formula:</b>	C5H11ClO2
<b>SMILES:</b>	CCOCC(O)CCl
<b>Mol. weight [g/mol]:</b>	138.59
<b>CAS:</b>	4151-98-8

## Physical Properties

Property code	Value	Unit	Source
gf	-264.97	kJ/mol	Joback Method
hf	-452.00	kJ/mol	Joback Method
hfus	14.66	kJ/mol	Joback Method
hvap	49.81	kJ/mol	Joback Method
log10ws	-0.53		Crippen Method
logp	0.623		Crippen Method
mcvol	105.290	ml/mol	McGowan Method
pc	3664.21	kPa	Joback Method
rinpol	960.00		NIST Webbook
tb	465.39	K	Joback Method
tc	637.15	K	Joback Method
tf	244.08	K	Joback Method
vc	0.396	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	215.28	J/molxK	465.39	Joback Method
cpg	253.53	J/molxK	608.52	Joback Method
cpg	246.44	J/molxK	579.89	Joback Method
cpg	239.08	J/molxK	551.27	Joback Method
cpg	231.43	J/molxK	522.64	Joback Method

cpg	223.50	J/mol×K	494.02	Joback Method
cpg	260.34	J/mol×K	637.15	Joback Method
dvisc	0.0001895	Paxs	465.39	Joback Method
dvisc	0.0003200	Paxs	428.50	Joback Method
dvisc	0.0005963	Paxs	391.62	Joback Method
dvisc	0.0012648	Paxs	354.74	Joback Method
dvisc	0.0031943	Paxs	317.85	Joback Method
dvisc	0.0102889	Paxs	280.97	Joback Method
dvisc	0.0471957	Paxs	244.08	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4151988&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4151988&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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