

# Bis(dimethylthiocarbamyl) sulfide

**Other names:** Tetramethylthiuram monosulfide  
Thiodicarbonic diamide ( $[(H_2-N)C(S)]_2-S$ ), tetramethyl-  
Aceto TMTM  
Bis(dimethylthiocarbamoyl) sulfide  
Bis(dimethylthiocarbamyl) monosulfide  
Carbamic acid, dimethyldithio-, anhydrosulfide  
Carbamic anhydride, tetramethyltrithio-  
Carbamodithioic acid, dimethyl-, anhydrosulfide  
Cyuram MS  
Ekagom TM  
Formamide, 1,1'-thiobis(N,N-dimethylthio-  
Monex  
Mono-Thiurad  
Monothiuram  
N,N,N',N'-Tetramethylthiuram monosulfide  
Sulfide, bis(dimethylthiocarbamoyl)  
Sulfide, bis((dimethylamino)thioxomethyl)  
Tetramethyldithiocarbamic acid anhydrosulfide  
Tetramethylthiuram sulfide  
Tetramethylthiuramide sulfide  
Thiuram monosulfide, tetramethyl-  
Thiuram MM  
TMTM  
TMTMS  
Unads  
USAF EK-P-6255  
USAF B-32  
Vulkacit Thiuram MS  
Monosulfure de tetramethylthiurame  
Pennac MS  
Tetramethylthiuramonosulfide  
Vulkacit MS  
Vulkacit thiuram ms/C  
Perkacit TMTM  
Thiodicarbonic diamide, tetramethyl-  
NSC 3400  
Tetramethylthiurammonium sulfide  
NSC 4767  
tetramethylthiuram monosulphide

**Inchi:** InChI=1S/C6H12N2S3/c1-7(2)5(9)11-6(10)8(3)4/h1-4H3

**InchiKey:** REQQFUJGGOFQL-UHFFFAOYSA-N  
**Formula:** C6H12N2S3  
**SMILES:** CN(C)C(=S)SC(=S)N(C)C  
**Mol. weight [g/mol]:** 208.37  
**CAS:** 97-74-5

## Physical Properties

Property code	Value	Unit	Source
chs	-5930.50 ± 1.00	kJ/mol	NIST Webbook
gf	488.44	kJ/mol	Joback Method
hf	302.76	kJ/mol	Joback Method
hfs	47.50 ± 1.30	kJ/mol	NIST Webbook
hfus	30.67	kJ/mol	Joback Method
hvap	53.31	kJ/mol	Joback Method
log10ws	-2.05		Crippen Method
logp	1.413		Crippen Method
mcvol	155.810	ml/mol	McGowan Method
pc	3920.94	kPa	Joback Method
tb	570.42	K	Joback Method
tc	808.21	K	Joback Method
tf	325.26	K	Joback Method
vc	0.533	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	331.73	J/mol×K	570.42	Joback Method
cpg	343.27	J/mol×K	610.05	Joback Method
cpg	353.79	J/mol×K	649.68	Joback Method
cpg	363.41	J/mol×K	689.31	Joback Method
cpg	372.25	J/mol×K	728.95	Joback Method
cpg	380.44	J/mol×K	768.58	Joback Method
cpg	388.08	J/mol×K	808.21	Joback Method
cps	261.50	J/mol×K	298.15	NIST Webbook

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C97745&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C97745&amp;Units=SI</a>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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