

# 2-Hexanone, 1,3-dichloro

<b>Inchi:</b>	InChI=1S/C6H10Cl2O/c1-2-3-5(8)6(9)4-7/h5H,2-4H2,1H3
<b>InchiKey:</b>	CICHFLZUYWZVLC-UHFFFAOYSA-N
<b>Formula:</b>	C6H10Cl2O
<b>SMILES:</b>	CCCC(Cl)C(=O)CCl
<b>Mol. weight [g/mol]:</b>	169.05

## Physical Properties

Property code	Value	Unit	Source
gf	-155.58	kJ/mol	Joback Method
hf	-316.51	kJ/mol	Joback Method
hfus	17.77	kJ/mol	Joback Method
hvap	44.08	kJ/mol	Joback Method
log10ws	-2.03		Crippen Method
logp	2.202		Crippen Method
mcvol	121.450	ml/mol	McGowan Method
pc	3096.73	kPa	Joback Method
rinpola	1018.00		NIST Webbook
rinpola	1018.00		NIST Webbook
tb	464.97	K	Joback Method
tc	660.74	K	Joback Method
tf	252.15	K	Joback Method
vc	0.469	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	227.76	J/molxK	464.97	Joback Method
cpg	237.47	J/molxK	497.60	Joback Method
cpg	246.69	J/molxK	530.23	Joback Method
cpg	255.45	J/molxK	562.86	Joback Method
cpg	263.76	J/molxK	595.48	Joback Method
cpg	271.63	J/molxK	628.11	Joback Method
cpg	279.08	J/molxK	660.74	Joback Method
dvisc	0.0053974	Paxs	252.15	Joback Method

dvisc	0.0025891	Paxs	287.62	Joback Method
dvisc	0.0014593	Paxs	323.09	Joback Method
dvisc	0.0009213	Paxs	358.56	Joback Method
dvisc	0.0006319	Paxs	394.03	Joback Method
dvisc	0.0004612	Paxs	429.50	Joback Method
dvisc	0.0003532	Paxs	464.97	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R629726&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R629726&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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