

# Benzoic acid, 4-methyl-3-nitro-

<b>Other names:</b>	p-Toluic acid, 3-nitro- 3-Nitro-4-methylbenzoic acid 4-Methyl-3-nitrobenzoic acid 3-Nitro-p-toluic acid 3-Nitro-para-toluic acid
<b>Inchi:</b>	InChI=1S/C8H7NO4/c1-5-2-3-6(8(10)11)4-7(5)9(12)13/h2-4H,1H3,(H,10,11)
<b>InchiKey:</b>	BBEWSMNRCUXQRF-UHFFFAOYSA-N
<b>Formula:</b>	C8H7NO4
<b>SMILES:</b>	<chem>Cc1ccc(C(=O)O)cc1[N+](=O)[O-]</chem>
<b>Mol. weight [g/mol]:</b>	181.15
<b>CAS:</b>	96-98-0

## Physical Properties

Property code	Value	Unit	Source
gf	-120.56	kJ/mol	Joback Method
hf	-270.43	kJ/mol	Joback Method
hfus	26.79	kJ/mol	Joback Method
hvap	77.02	kJ/mol	Joback Method
log10ws	-2.66		Crippen Method
logp	1.601		Crippen Method
mcvol	124.680	ml/mol	McGowan Method
pc	4339.67	kPa	Joback Method
tb	716.97	K	Joback Method
tc	947.46	K	Joback Method
tf	485.74	K	Joback Method
vc	0.482	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	314.41	J/mol×K	716.97	Joback Method
cpg	322.66	J/mol×K	755.38	Joback Method
cpg	330.28	J/mol×K	793.80	Joback Method
cpg	337.28	J/mol×K	832.21	Joback Method

cpg	343.70	J/mol×K	870.63	Joback Method
cpg	349.56	J/mol×K	909.04	Joback Method
cpg	354.89	J/mol×K	947.46	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C96980&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C96980&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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