

N-Acetyl-D-glucosamine

Other names:

2-Acetamido-2-deoxy-D-glucose
2-Acetamido-2-deoxyglucose
2-Acetamido-D-glucose
2-acetylamino-2-deoxy-D-glucose
Acetylglucosamine
D-Glucose, 2-(acetylamino)-2-deoxy-
D-Glucose, 2-acetamido-2-deoxy-
N-((2R,3R,4S,5R)-3,4,5,6-tetrahydroxy-1-oxohexan-2-yl)acetamide
N-Acetylglucosamine
N-acetyl-«beta»-D-glucosamine
NSC 524344
d-N-Acetylglucosamine

Inchi:

InChI=1S/C8H15NO6/c1-4(12)9-5(2-10)7(14)8(15)6(13)3-11/h2,5-8,11,13-15H,3H2,1H3,

InchiKey:

MBLBDJOUHNCFQT-UHFFFAOYSA-N

Formula:

C8H15NO6

SMILES:

CC(O)=NC(C=O)C(O)C(O)C(O)CO

Mol. weight [g/mol]:

221.21

CAS:

7512-17-6

Physical Properties

Property code	Value	Unit	Source
hf	-1003.87	kJ/mol	Joback Method
hvap	125.36	kJ/mol	Joback Method
log10ws	1.06		Crippen Method
logp	-2.395		Crippen Method
mcvol	160.180	ml/mol	McGowan Method
pc	4328.25	kPa	Joback Method
tb	966.80	K	Joback Method
tc	1189.18	K	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cps	269.80	J/mol×K	288.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	276.40	J/mol×K	293.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	276.90	J/mol×K	298.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	278.00	J/mol×K	303.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	283.30	J/mol×K	308.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	281.00	J/mol×K	313.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	281.40	J/mol×K	318.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	292.70	J/mol×K	323.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	293.10	J/mol×K	328.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides

cps	299.40	J/molxK	333.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	303.80	J/molxK	338.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	308.70	J/molxK	343.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	319.70	J/molxK	348.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	325.20	J/molxK	353.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides
cps	332.00	J/molxK	358.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides

Sources

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C7512176&Units=SI>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Apparent molar volumes and apparent molar heat capacities of aqueous

<https://www.doi.org/10.1016/j.jct.2006.04.007>

Temperature dependence of the heat capacities in the solid state of 18

<https://www.doi.org/10.1016/j.jct.2008.08.007>

mono-, di-, and poly-saccharides at 278.15 K and for aqueous trimethylacetamide

https://en.wikipedia.org/wiki/Joback_method

temperatures from 278.15 K to 393.15 K at the pressure 0.35 MPa:

<http://link.springer.com/article/10.1007/BF02311772>

McCowan Method:

Legend

cps:	Solid phase heat capacity
hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature

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