

# Cyclohexanecarboxylic acid, 2,2,2-trichloroethyl ester

**Inchi:** InChI=1S/C9H13Cl3O2/c10-9(11,12)6-14-8(13)7-4-2-1-3-5-7/h7H,1-6H2  
**InchiKey:** LDNCHQISWJTDHK-UHFFFAOYSA-N  
**Formula:** C9H13Cl3O2  
**SMILES:** O=C(OCC(Cl)(Cl)Cl)C1CCCCC1  
**Mol. weight [g/mol]:** 259.56

## Physical Properties

Property code	Value	Unit	Source
gf	-217.52	kJ/mol	Joback Method
hf	-475.54	kJ/mol	Joback Method
hfus	18.86	kJ/mol	Joback Method
hvap	57.07	kJ/mol	Joback Method
log10ws	-3.67		Crippen Method
logp	3.480		Crippen Method
mvol	170.970	ml/mol	McGowan Method
pc	2704.22	kPa	Joback Method
rinpol	1563.00		NIST Webbook
rinpol	1563.00		NIST Webbook
tb	610.22	K	Joback Method
tc	845.54	K	Joback Method
tf	362.91	K	Joback Method
vc	0.632	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	398.15	J/molxK	610.22	Joback Method
cpg	462.45	J/molxK	806.32	Joback Method
cpg	451.70	J/molxK	767.10	Joback Method
cpg	439.95	J/molxK	727.88	Joback Method
cpg	427.14	J/molxK	688.66	Joback Method
cpg	413.22	J/molxK	649.44	Joback Method
cpg	472.24	J/molxK	845.54	Joback Method
dvisc	0.0002013	Paxs	610.22	Joback Method

dvisc	0.0002678	Paxs	569.00	Joback Method
dvisc	0.0003725	Paxs	527.78	Joback Method
dvisc	0.0005479	Paxs	486.57	Joback Method
dvisc	0.0008658	Paxs	445.35	Joback Method
dvisc	0.0015018	Paxs	404.13	Joback Method
dvisc	0.0029522	Paxs	362.91	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U354650&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U354650&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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