

Succinic acid, 2,2-dichloroethyl 1,1,1-trifluoroprop-2-yl ester

Inchi:	InChI=1S/C9H11Cl2F3O4/c1-5(9(12,13)14)18-8(16)3-2-7(15)17-4-6(10)11/h5-6H,2-4H2,
InchiKey:	MPEONHWHNJRGO-UHFFFAOYSA-N
Formula:	C9H11Cl2F3O4
SMILES:	CC(OC(=O)CCC(=O)OCC(Cl)Cl)C(F)(F)F
Mol. weight [g/mol]:	311.08

Physical Properties

Property code	Value	Unit	Source
gf	-1053.27	kJ/mol	Joback Method
hf	-1357.81	kJ/mol	Joback Method
hfus	27.81	kJ/mol	Joback Method
hvap	58.19	kJ/mol	Joback Method
log10ws	-3.00		Crippen Method
logp	2.608		Crippen Method
mvol	182.340	ml/mol	McGowan Method
pc	2135.43	kPa	Joback Method
rinpol	1421.00		NIST Webbook
rinpol	1421.00		NIST Webbook
tb	626.46	K	Joback Method
tc	809.99	K	Joback Method
tf	369.54	K	Joback Method
vc	0.717	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	449.21	J/mol×K	626.46	Joback Method
cpg	459.77	J/mol×K	657.05	Joback Method
cpg	469.71	J/mol×K	687.64	Joback Method
cpg	479.05	J/mol×K	718.22	Joback Method
cpg	487.80	J/mol×K	748.81	Joback Method
cpg	495.98	J/mol×K	779.40	Joback Method
cpg	503.59	J/mol×K	809.99	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U390842&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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