

# Terephthalic acid, 2-iodobenzyl propyl ester

<b>Inchi:</b>	InChI=1S/C18H17IO4/c1-2-11-22-17(20)13-7-9-14(10-8-13)18(21)23-12-15-5-3-4-6-16(1
<b>InchiKey:</b>	QPSZNHNNQELKO-UHFFFAOYSA-N
<b>Formula:</b>	C18H17IO4
<b>SMILES:</b>	CCCOC(=O)c1ccc(C(=O)OCc2ccccc2I)cc1
<b>Mol. weight [g/mol]:</b>	424.23

## Physical Properties

Property code	Value	Unit	Source
gf	-103.48	kJ/mol	Joback Method
hf	-377.46	kJ/mol	Joback Method
hfus	39.66	kJ/mol	Joback Method
hvap	89.22	kJ/mol	Joback Method
log10ws	-6.05		Crippen Method
logp	4.215		Crippen Method
mvol	257.660	ml/mol	McGowan Method
pc	1985.89	kPa	Joback Method
rinpol	2970.00		NIST Webbook
rinpol	2970.00		NIST Webbook
tb	920.28	K	Joback Method
tc	1169.28	K	Joback Method
tf	572.88	K	Joback Method
vc	0.964	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	708.27	J/molxK	920.28	Joback Method
cpg	753.04	J/molxK	1127.78	Joback Method
cpg	746.43	J/molxK	1086.28	Joback Method
cpg	738.70	J/molxK	1044.78	Joback Method
cpg	729.80	J/molxK	1003.28	Joback Method
cpg	719.67	J/molxK	961.78	Joback Method
cpg	758.57	J/molxK	1169.28	Joback Method
dvisc	0.0000552	Paxs	920.28	Joback Method

dvisc	0.0000690	Paxs	862.38	Joback Method
dvisc	0.0000891	Paxs	804.48	Joback Method
dvisc	0.0001197	Paxs	746.58	Joback Method
dvisc	0.0001689	Paxs	688.68	Joback Method
dvisc	0.0002540	Paxs	630.78	Joback Method
dvisc	0.0004149	Paxs	572.88	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U416072&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U416072&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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