

# Palmidrol

<b>Other names:</b>	Hexadecanamide, N-(2-hydroxyethyl)- Hydroxyethylpalmitamide Impulsin Loramine P 256 N-(2-Hydroxyethyl)palmitamide Palmitic acid monoethanolamide Palmitoylethanolamide N-(2-Hydroxyethyl)hexadecanamide NSC 23320
<b>Inchi:</b>	InChI=1S/C18H37NO2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-18(21)19-16-17-20/h20H,2
<b>InchiKey:</b>	HXYVTAGFYLMHSO-UHFFFAOYSA-N
<b>Formula:</b>	C18H37NO2
<b>SMILES:</b>	CCCCCCCCCCCCCCCC(O)=NCCO
<b>Mol. weight [g/mol]:</b>	299.49
<b>CAS:</b>	544-31-0

## Physical Properties

Property code	Value	Unit	Source
hf	-646.88	kJ/mol	Joback Method
hvap	92.41	kJ/mol	Joback Method
log10ws	-5.60		Crippen Method
logp	5.417		Crippen Method
mvol	281.900	ml/mol	McGowan Method
pc	1220.85	kPa	Joback Method
rinpol	2140.50		NIST Webbook
rinpol	2140.50		NIST Webbook
tb	872.16	K	Joback Method
tc	1068.26	K	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C544310&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C544310&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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