

3'-methoxy,4'-hydroxyphenyl-ethyl)-1,2,3,4-tetrahydromethylquinoline

Inchi: InChI=1S/C18H21NO2/c1-21-18-12-13(7-11-17(18)20)6-9-15-10-8-14-4-2-3-5-16(14)19-
InchiKey: SVRGVWFUWFSROF-UHFFFAOYSA-N
Formula: C18H21NO2
SMILES: COc1cc(CCC2CCc3ccccc3N2)ccc1O
Mol. weight [g/mol]: 283.36

Physical Properties

Property code	Value	Unit	Source
gf	182.98	kJ/mol	Joback Method
hf	-169.81	kJ/mol	Joback Method
hfus	42.28	kJ/mol	Joback Method
hvap	83.80	kJ/mol	Joback Method
log10ws	-4.41		Crippen Method
logp	3.760		Crippen Method
mcvol	227.820	ml/mol	McGowan Method
pc	2495.01	kPa	Joback Method
rinpol	2676.00		NIST Webbook
tb	837.16	K	Joback Method
tc	1086.83	K	Joback Method
tf	623.90	K	Joback Method
vc	0.797	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	698.67	J/mol×K	837.16	Joback Method
cpg	715.35	J/mol×K	878.77	Joback Method
cpg	730.98	J/mol×K	920.38	Joback Method
cpg	745.71	J/mol×K	961.99	Joback Method
cpg	759.67	J/mol×K	1003.60	Joback Method
cpg	773.01	J/mol×K	1045.21	Joback Method
cpg	785.85	J/mol×K	1086.83	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R398308&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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