

# Succinic acid, hexyl 2-methoxy-4-chlorobenzyl ester

<b>Inchi:</b>	InChI=1S/C18H25ClO5/c1-3-4-5-6-11-23-17(20)9-10-18(21)24-13-14-7-8-15(19)12-16(14)
<b>InchiKey:</b>	FKYULHQRAMXWRF-UHFFFAOYSA-N
<b>Formula:</b>	C18H25ClO5
<b>SMILES:</b>	CCCCCOC(=O)CCC(=O)OCc1ccc(Cl)cc1OC
<b>Mol. weight [g/mol]:</b>	356.84

## Physical Properties

Property code	Value	Unit	Source
gf	-390.94	kJ/mol	Joback Method
hf	-838.82	kJ/mol	Joback Method
hfus	46.60	kJ/mol	Joback Method
hvap	84.37	kJ/mol	Joback Method
log10ws	-5.07		Crippen Method
logp	4.296		Crippen Method
mvol	273.710	ml/mol	McGowan Method
pc	1471.36	kPa	Joback Method
rinpol	2533.00		NIST Webbook
rinpol	2533.00		NIST Webbook
tb	860.31	K	Joback Method
tc	1065.97	K	Joback Method
tf	540.55	K	Joback Method
vc	1.050	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	815.46	J/mol×K	860.31	Joback Method
cpg	829.42	J/mol×K	894.59	Joback Method
cpg	842.23	J/mol×K	928.86	Joback Method
cpg	853.90	J/mol×K	963.14	Joback Method
cpg	864.42	J/mol×K	997.42	Joback Method
cpg	873.80	J/mol×K	1031.70	Joback Method
cpg	882.04	J/mol×K	1065.97	Joback Method
dvisc	0.0003673	Paxs	540.55	Joback Method

dvisc	0.0002245	Paxs	593.84	Joback Method
dvisc	0.0001488	Paxs	647.14	Joback Method
dvisc	0.0001050	Paxs	700.43	Joback Method
dvisc	0.0000778	Paxs	753.72	Joback Method
dvisc	0.0000600	Paxs	807.02	Joback Method
dvisc	0.0000478	Paxs	860.31	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U380850&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U380850&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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