

# 2,4,6 Tri-tert-butylNitrobenzene

<b>Other names:</b>	2,4,6-tri-t-ButylNitrobenzene Benzene, 1,3,5-tris(1,1-dimethylethyl)-2-nitro- Nitrobenzene, 2,4,6-tri-tert-butyl- 1,3,5-tri(tert-butyl)-2-nitrobenzene
<b>Inchi:</b>	InChI=1S/C18H29NO2/c1-16(2,3)12-10-13(17(4,5)6)15(19(20)21)14(11-12)18(7,8)9/h10
<b>InchiKey:</b>	IMDZOFHRUMJNQR-UHFFFAOYSA-N
<b>Formula:</b>	C18H29NO2
<b>SMILES:</b>	<chem>CC(C)(C)c1cc(C(C)(C)C)c([N+](=O)[O-])c(C(C)(C)C)c1</chem>
<b>Mol. weight [g/mol]:</b>	291.43
<b>CAS:</b>	4074-25-3

## Physical Properties

Property code	Value	Unit	Source
gf	228.27	kJ/mol	Joback Method
hf	-249.74	kJ/mol	Joback Method
hfus	24.37	kJ/mol	Joback Method
hvap	72.63	kJ/mol	Joback Method
log10ws	-6.06		Crippen Method
logp	5.487		Crippen Method
mcvol	258.140	ml/mol	McGowan Method
pc	1489.58	kPa	Joback Method
tb	795.01	K	Joback Method
tc	1033.56	K	Joback Method
tf	507.47	K	Joback Method
vc	0.985	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	788.47	J/molxK	795.01	Joback Method
cpg	806.80	J/molxK	834.77	Joback Method
cpg	823.86	J/molxK	874.53	Joback Method
cpg	839.78	J/molxK	914.29	Joback Method
cpg	854.72	J/molxK	954.04	Joback Method

cpg	868.82	J/mol×K	993.80	Joback Method
cpg	882.23	J/mol×K	1033.56	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4074253&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4074253&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcpol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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