

Fumaric acid, 2,2,2-trichloroethyl tridecyl ester

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|-----------------------------|---|
| Inchi: | InChI=1S/C19H31Cl3O4/c1-2-3-4-5-6-7-8-9-10-11-12-15-25-17(23)13-14-18(24)26-16-19 |
| InchiKey: | ZTXCYTUNNUJFBB-BUHFOSPRSA-N |
| Formula: | C19H31Cl3O4 |
| SMILES: | CCCCCCCCCCCCOC(=O)C=CC(=O)OCC(Cl)(Cl)Cl |
| Mol. weight [g/mol]: | 429.81 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -311.47 | kJ/mol | Joback Method |
| hf | -863.84 | kJ/mol | Joback Method |
| hfus | 55.92 | kJ/mol | Joback Method |
| hvap | 88.02 | kJ/mol | Joback Method |
| log10ws | -6.92 | | Crippen Method |
| logp | 6.310 | | Crippen Method |
| mvol | 325.870 | ml/mol | McGowan Method |
| pc | 1118.56 | kPa | Joback Method |
| rinpol | 2684.00 | | NIST Webbook |
| rinpol | 2684.00 | | NIST Webbook |
| tb | 899.92 | K | Joback Method |
| tc | 1105.02 | K | Joback Method |
| tf | 535.31 | K | Joback Method |
| vc | 1.264 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 970.48 | J/molxK | 899.92 | Joback Method |
| cpg | 1033.95 | J/molxK | 1070.83 | Joback Method |
| cpg | 1022.96 | J/molxK | 1036.65 | Joback Method |
| cpg | 1011.18 | J/molxK | 1002.47 | Joback Method |
| cpg | 998.54 | J/molxK | 968.29 | Joback Method |
| cpg | 984.99 | J/molxK | 934.10 | Joback Method |
| cpg | 1044.20 | J/molxK | 1105.02 | Joback Method |
| dvisc | 0.0000256 | Paxs | 899.92 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0000341 | Paxs | 839.15 | Joback Method |
| dvisc | 0.0000476 | Paxs | 778.38 | Joback Method |
| dvisc | 0.0000701 | Paxs | 717.62 | Joback Method |
| dvisc | 0.0001111 | Paxs | 656.85 | Joback Method |
| dvisc | 0.0001932 | Paxs | 596.08 | Joback Method |
| dvisc | 0.0003810 | Paxs | 535.31 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U348510&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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<https://www.chemeo.com/cid/97-690-9/Fumaric-acid-2-2-2-trichloroethyl-tridecyl-ester.pdf>

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