

Cyanoacetic acid, hexadecyl ester

Inchi:

InchiKey:

Formula:

SMILES:

Mol. weight [g/mol]:

InChI=1S/C19H35NO2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-18-22-19(21)16-17-20/h2-

VYKJKXWAMIAGRC-UHFFFAOYSA-N

C19H35NO2

CCCCCCCCCCCCCCCCOC(=O)CC#N

309.49

Physical Properties

Property code	Value	Unit	Source
gf	8.36	kJ/mol	Joback Method
hf	-515.41	kJ/mol	Joback Method
hfus	49.26	kJ/mol	Joback Method
hvap	77.52	kJ/mol	Joback Method
log10ws	-6.50		Crippen Method
logp	5.925		Crippen Method
mcvol	287.390	ml/mol	McGowan Method
pc	1095.72	kPa	Joback Method
rinpol	2357.00		NIST Webbook
rinpol	2357.00		NIST Webbook
tb	812.49	K	Joback Method
tc	999.36	K	Joback Method
tf	441.04	K	Joback Method
vc	1.149	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	895.17	J/mol×K	812.49	Joback Method
cpg	912.11	J/mol×K	843.63	Joback Method
cpg	928.12	J/mol×K	874.78	Joback Method
cpg	943.22	J/mol×K	905.92	Joback Method
cpg	957.43	J/mol×K	937.07	Joback Method
cpg	970.78	J/mol×K	968.21	Joback Method
cpg	983.31	J/mol×K	999.36	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U406233&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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