

Diglycolic acid, 2-chloro-6-fluorophenyl nonyl ester

Inchi:	InChI=1S/C19H26ClFO5/c1-2-3-4-5-6-7-8-12-25-17(22)13-24-14-18(23)26-19-15(20)10-9
InchiKey:	HFKQIFMVKMPXNE-UHFFFAOYSA-N
Formula:	C19H26ClFO5
SMILES:	CCCCCCCCCOC(=O)COCC(=O)Oc1c(F)cccc1Cl
Mol. weight [g/mol]:	388.86

Physical Properties

Property code	Value	Unit	Source
gf	-577.33	kJ/mol	Joback Method
hf	-1055.57	kJ/mol	Joback Method
hfus	52.27	kJ/mol	Joback Method
hvap	85.78	kJ/mol	Joback Method
log10ws	-5.36		Crippen Method
logp	4.695		Crippen Method
mvol	289.570	ml/mol	McGowan Method
pc	1324.24	kPa	Joback Method
rinpol	3272.00		NIST Webbook
rinpol	3272.00		NIST Webbook
tb	882.46	K	Joback Method
tc	1085.81	K	Joback Method
tf	552.41	K	Joback Method
vc	1.125	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	880.91	J/mol×K	882.46	Joback Method
cpg	894.73	J/mol×K	916.35	Joback Method
cpg	907.38	J/mol×K	950.24	Joback Method
cpg	918.86	J/mol×K	984.13	Joback Method
cpg	929.17	J/mol×K	1018.02	Joback Method
cpg	938.33	J/mol×K	1051.92	Joback Method
cpg	946.33	J/mol×K	1085.81	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U381947&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpolar:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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