

# Succinic acid, 5-bromo-2-methoxybenzyl 2-methylhex-3-yl ester

<b>Inchi:</b>	InChI=1S/C19H27BrO5/c1-5-6-16(13(2)3)25-19(22)10-9-18(21)24-12-14-11-15(20)7-8-1
<b>InchiKey:</b>	ANVGYEQLPHJOIC-UHFFFAOYSA-N
<b>Formula:</b>	C19H27BrO5
<b>SMILES:</b>	CCCC(OC(=O)CCC(=O)OCc1cc(Br)ccc1OC)C(C)C
<b>Mol. weight [g/mol]:</b>	415.32

## Physical Properties

Property code	Value	Unit	Source
gf	-361.15	kJ/mol	Joback Method
hf	-827.95	kJ/mol	Joback Method
hfus	43.23	kJ/mol	Joback Method
hvap	87.87	kJ/mol	Joback Method
log10ws	-5.83		Crippen Method
logp	4.649		Crippen Method
mcvol	293.060	ml/mol	McGowan Method
pc	1501.15	kPa	Joback Method
rinpol	2602.00		NIST Webbook
rinpol	2602.00		NIST Webbook
tb	911.04	K	Joback Method
tc	1126.94	K	Joback Method
tf	551.70	K	Joback Method
vc	1.107	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	887.11	J/molxK	911.04	Joback Method
cpg	900.81	J/molxK	947.02	Joback Method
cpg	913.22	J/molxK	983.01	Joback Method
cpg	924.35	J/molxK	1018.99	Joback Method
cpg	934.22	J/molxK	1054.98	Joback Method
cpg	942.84	J/molxK	1090.96	Joback Method
cpg	950.22	J/molxK	1126.94	Joback Method
dvisc	0.0003204	Paxs	551.70	Joback Method

dvisc	0.0001800	Paxs	611.59	Joback Method
dvisc	0.0001120	Paxs	671.48	Joback Method
dvisc	0.0000754	Paxs	731.37	Joback Method
dvisc	0.0000539	Paxs	791.26	Joback Method
dvisc	0.0000403	Paxs	851.15	Joback Method
dvisc	0.0000314	Paxs	911.04	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U381077&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U381077&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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