

# Phthalic acid, 5-bromo-2-methoxybenzyl butyl ester

Inchi:	InChI=1S/C20H21BrO5/c1-3-4-11-25-19(22)16-7-5-6-8-17(16)20(23)26-13-14-12-15(21)
InchiKey:	KZJIFANCWXWHBU-UHFFFAOYSA-N
Formula:	C20H21BrO5
SMILES:	CCCCOC(=O)c1ccccc1C(=O)OCc1cc(Br)ccc1OC
Mol. weight [g/mol]:	421.28

## Physical Properties

Property code	Value	Unit	Source
gf	-245.07	kJ/mol	Joback Method
hf	-612.97	kJ/mol	Joback Method
hfus	46.52	kJ/mol	Joback Method
hvap	93.81	kJ/mol	Joback Method
log10ws	-6.60		Crippen Method
logp	4.772		Crippen Method
mvol	283.390	ml/mol	McGowan Method
pc	1769.87	kPa	Joback Method
rinpol	2799.00		NIST Webbook
rinpol	2799.00		NIST Webbook
tb	966.46	K	Joback Method
tc	1201.04	K	Joback Method
tf	631.91	K	Joback Method
vc	1.067	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	837.96	J/molxK	966.46	Joback Method
cpg	849.08	J/molxK	1005.56	Joback Method
cpg	858.78	J/molxK	1044.65	Joback Method
cpg	867.10	J/molxK	1083.75	Joback Method
cpg	874.04	J/molxK	1122.85	Joback Method
cpg	879.63	J/molxK	1161.95	Joback Method
cpg	883.89	J/molxK	1201.04	Joback Method
dvisc	0.0002036	Paxs	631.91	Joback Method

dvisc	0.0001342	Paxs	687.67	Joback Method
dvisc	0.0000941	Paxs	743.43	Joback Method
dvisc	0.0000694	Paxs	799.19	Joback Method
dvisc	0.0000532	Paxs	854.94	Joback Method
dvisc	0.0000422	Paxs	910.70	Joback Method
dvisc	0.0000343	Paxs	966.46	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382866&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382866&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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