

Benzenesulfonamide, 4-methyl-N-ethyl-N-undecyl-

Inchi: InChI=1S/C20H35NO2S/c1-4-6-7-8-9-10-11-12-13-18-21(5-2)24(22,23)20-16-14-19(3)15
InchiKey: CBQXCZMZRXXWGCY-UHFFFAOYSA-N
Formula: C20H35NO2S
SMILES: CCCCCCCCCCN(CC)S(=O)(=O)c1ccc(C)cc1
Mol. weight [g/mol]: 353.56

Physical Properties

Property code	Value	Unit	Source
gf	-137.46	kJ/mol	Joback Method
hf	-616.89	kJ/mol	Joback Method
hfus	55.61	kJ/mol	Joback Method
hvap	83.73	kJ/mol	Joback Method
log10ws	-6.22		Crippen Method
logp	5.536		Crippen Method
mcvol	306.970	ml/mol	McGowan Method
pc	1403.79	kPa	Joback Method
rinpol	2991.00		NIST Webbook
rinpol	2991.00		NIST Webbook
tb	748.88	K	Joback Method
tc	931.53	K	Joback Method
tf	425.13	K	Joback Method
vc	1.192	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	909.29	J/mol×K	748.88	Joback Method
cpg	928.56	J/mol×K	779.32	Joback Method
cpg	946.70	J/mol×K	809.76	Joback Method
cpg	963.76	J/mol×K	840.21	Joback Method
cpg	979.76	J/mol×K	870.65	Joback Method
cpg	994.75	J/mol×K	901.09	Joback Method
cpg	1008.76	J/mol×K	931.53	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U415281&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinsol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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