

# Phthalic acid, 2-(3-chlorophenyl)ethyl dodecyl ester

Inchi:	InChI=1S/C28H37ClO4/c1-2-3-4-5-6-7-8-9-10-13-20-32-27(30)25-17-11-12-18-26(25)28
InchiKey:	ZZHRZGJDOMXEBO-UHFFFAOYSA-N
Formula:	C28H37ClO4
SMILES:	CCCCCCCCCCCCOC(=O)c1ccccc1C(=O)OCCc1ccc(Cl)c1
Mol. weight [g/mol]:	473.04

## Physical Properties

Property code	Value	Unit	Source
gf	-89.33	kJ/mol	Joback Method
hf	-676.47	kJ/mol	Joback Method
hfus	65.35	kJ/mol	Joback Method
hvap	106.50	kJ/mol	Joback Method
log10ws	-9.28		Crippen Method
logp	7.817		Crippen Method
mvol	384.980	ml/mol	McGowan Method
pc	965.07	kPa	Joback Method
rinpol	3509.00		NIST Webbook
rinpol	3509.00		NIST Webbook
tb	1093.37	K	Joback Method
tc	1339.26	K	Joback Method
tf	657.44	K	Joback Method
vc	1.484	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1280.99	J/molxK	1093.37	Joback Method
cpg	1294.29	J/molxK	1134.35	Joback Method
cpg	1305.94	J/molxK	1175.33	Joback Method
cpg	1316.02	J/molxK	1216.32	Joback Method
cpg	1324.61	J/molxK	1257.30	Joback Method
cpg	1331.80	J/molxK	1298.28	Joback Method
cpg	1337.67	J/molxK	1339.26	Joback Method
dvisc	0.0001517	Paxs	657.44	Joback Method

dvisc	0.0000851	Paxs	730.10	Joback Method
dvisc	0.0000530	Paxs	802.75	Joback Method
dvisc	0.0000357	Paxs	875.40	Joback Method
dvisc	0.0000256	Paxs	948.06	Joback Method
dvisc	0.0000192	Paxs	1020.71	Joback Method
dvisc	0.0000150	Paxs	1093.37	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U377862&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U377862&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>c</sub>vol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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