

# 3-methyl-1,2-dithiolan-4-one

<b>Other names:</b>	1,2-Dithiolan-4-one, 3-methyl
<b>Inchi:</b>	InChI=1S/C4H6OS2/c1-3-4(5)2-6-7-3/h3H,2H2,1H3
<b>InchiKey:</b>	ARWPSTDFVNPXSC-UHFFFAOYSA-N
<b>Formula:</b>	C4H6OS2
<b>SMILES:</b>	CC1SSCC1=O
<b>Mol. weight [g/mol]:</b>	134.22

## Physical Properties

Property code	Value	Unit	Source
gf	-23.52	kJ/mol	Joback Method
hf	-112.59	kJ/mol	Joback Method
hfus	6.88	kJ/mol	Joback Method
hvap	40.63	kJ/mol	Joback Method
log10ws	-1.54		Crippen Method
logp	1.339		Crippen Method
mcvol	90.630	ml/mol	McGowan Method
pc	5109.34	kPa	Joback Method
rinpol	1071.00		NIST Webbook
rinpol	1106.00		NIST Webbook
rinpol	1090.00		NIST Webbook
rinpol	1071.00		NIST Webbook
rinpol	1071.00		NIST Webbook
rinpol	1072.00		NIST Webbook
rinpol	1071.00		NIST Webbook
rinpol	1102.00		NIST Webbook
rinpol	1103.00		NIST Webbook
rinpol	1090.00		NIST Webbook
rinpol	1072.00		NIST Webbook
rinpol	1098.00		NIST Webbook
rinpol	1103.00		NIST Webbook
ripol	1655.00		NIST Webbook
ripol	1664.00		NIST Webbook
ripol	1664.00		NIST Webbook
ripol	1655.00		NIST Webbook
tb	469.68	K	Joback Method
tc	727.72	K	Joback Method
tf	380.86	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	168.16	J/mol×K	469.68	Joback Method
cpg	178.88	J/mol×K	512.69	Joback Method
cpg	189.07	J/mol×K	555.69	Joback Method
cpg	198.71	J/mol×K	598.70	Joback Method
cpg	207.80	J/mol×K	641.71	Joback Method
cpg	216.33	J/mol×K	684.72	Joback Method
cpg	224.29	J/mol×K	727.72	Joback Method

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=R618296&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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